

ICP Construction Inc.

Version No: 3.6

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

SECTION 1 Identification

Product Identifier

Product name	HandiFoam HFO FR Black B-side	
Synonyms	Not Available	
Proper shipping name	Chemical under pressure, n.o.s.(Contains Hydrofluoroolefin, Nitrogen)	
Chemical formula	Not Applicable	
Other means of identification	Not Available	

Recommended use of the chemical and restrictions on use

Relevant identified uses Low pressure polyurethane foam, Side-B Component, for PROFESSIONAL USE ONLY

Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	ICP Construction Inc.	
Address	0 Dascomb Road Andover, MA 01810 United States	
Telephone	1-866-667-5119 1-978-623-9987	
Fax	Not Available	
Website	www.icpgroup.com	
Email	sds@icpgroup.com	

Emergency phone number

0 71	
Association / Organisation	ChemTel
Emergency telephone numbers	1-800-255-3924
Other emergency telephone numbers	1-813-248-0585

SECTION 2 Hazard(s) identification

Classification of the substance or mixture

NFPA 704 diamond

2 1	of this S Health F	e hazard category numbers found in GHS classification in section 2 DSs are NOT to be used to fill in the NFPA 704 diamond. Blue = ted = Fire Yellow = Reactivity White = Special (Oxidizer or water substances)
Classif	ication	Serious Eye Damage/Eye Irritation Category 2A, Skin Corrosion/Irritation Category 2, Reproductive Toxicity Category 1B, Gases Under Pressure (Compressed Gas), Sensitisation (Skin) Category 1
_abel elements		
_abel elements Hazard pictog	gram(s)	

Hazard statement(s)

H319 Causes serious eye irritation.

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H315	Causes skin irritation.
H360	May damage fertility or the unborn child.
H280	Contains gas under pressure; may explode if heated.
H317	May cause an allergic skin reaction.

Hazard(s) not otherwise classified

Not Applicable

Precautionary statement(s) Prevention

P201	Obtain special instructions before use.	
P280	ar protective gloves, protective clothing, eye protection and face protection.	
P261	Avoid breathing gas.	
P202	Do not handle until all safety precautions have been read and understood.	
P264	Wash all exposed external body areas thoroughly after handling.	
P272	Contaminated work clothing must not be allowed out of the workplace.	

Precautionary statement(s) Response

P308+P313	IF exposed or concerned: Get medical advice/ attention.	
P305+P351+P338	IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.	
P333+P313	If skin irritation or rash occurs: Get medical advice/attention.	
P337+P313	If eye irritation persists: Get medical advice/attention.	
P302+P352	IF ON SKIN: Wash with plenty of water and soap.	
P332+P313	If skin irritation occurs: Get medical advice/attention.	
P362+P364	Take off contaminated clothing and wash it before reuse.	

Precautionary statement(s) Storage

P405	Store locked up.	
P410+P403	Protect from sunlight. Store in a well-ventilated place.	

Precautionary statement(s) Disposal

Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

SECTION 3 Composition / information on ingredients

P501

Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
111-46-6	5-10	diethylene glycol
13674-84-5*	15-40	tris(2-chloroisopropyl)phosphate
1185-81-5	0.1-1	dibutyltin dilauryl mercaptide
107-21-1	0.1-0.5	ethylene glycol
7560-83-0	<1	N-methyldicyclohexylamine
29118-24-9	10-30	1.3.3.3-tetrafluoropropene
Not Available	<5	Nitrogen

The specific chemical identity and/or exact percentage (concentration) of composition has been withheld as a trade secret.

SECTION 4 First-aid measures

Description of first aid measures

Eye Contact	 If product comes in contact with eyes remove the patient from gas source or contaminated area. Take the patient to the nearest eye wash, shower or other source of clean water. Open the eyelid(s) wide to allow the material to evaporate. Gently rinse the affected eye(s) with clean, cool water for at least 15 minutes. Have the patient lie or sit down and tilt the head back. Hold the eyelid(s) open and pour water slowly over the eyeball(s) at the inner corners, letting the water run out of the outer corners. The patient may be in great pain and wish to keep the eyes closed. It is important that the material is rinsed from the eyes to prevent further damage. Ensure that the patient looks up, and side to side as the eye is rinsed in order to better reach all parts of the eye(s) Transport to hospital or doctor. Even when no pain persists and vision is good, a doctor should examine the eye as delayed damage may occur. If the patient contolerate light, protect the eyes with a clean, loosely tied bandage. Ensure verbal communication and physical contact with the patient. DO NOT allow the patient to tightly shut the eyes DO NOT introduce oil or ointment into the eye(s) without medical advice
	DO NOT introduce oil or ointment into the eye(s) without medical advice DO NOT use hot or tepid water.

Skin Contact	 If skin or hair contact occurs: Immediately flush body and clothes with large amounts of water, using safety shower if available. Quickly remove all contaminated clothing, including footwear. Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre. Transport to hospital, or doctor.
Inhalation	 Following exposure to gas, remove the patient from the gas source or contaminated area. NOTE: Personal Protective Equipment (PPE), including positive pressure self-contained breathing apparatus may be required to assure the safety of the rescuer. Prostheses such as false teeth, which may block the airway, should be removed, where possible, prior to initiating first aid procedures. If the patient is not breathing spontaneously, administer rescue breathing. If the patient does not have a pulse, administer CPR. If medical oxygen and appropriately trained personnel are available, administer 100% oxygen. Summon an emergency ambulance. If an ambulance is not available, contact a physician, hospital, or Poison Control Centre for further instruction. Keep the patient warm, comfortable and at rest while awaiting medical care. MONITOR THE BREATHING AND PULSE, CONTINUOUSLY. Administer rescue breathing (preferably with a demand-valve resuscitator, bag-valve mask-device, or pocket mask as trained) or CPR if necessary.
Ingestion	Not considered a normal route of entry. Avoid giving milk or oils. Avoid giving alcohol.

Most important symptoms and effects, both acute and delayed

See Section 11

Indication of any immediate medical attention and special treatment needed

for intoxication due to Freons/ Halons;

- A: Emergency and Supportive Measures
- Maintain an open airway and assist ventilation if necessary
- Treat coma and arrhythmias if they occur. Avoid (adrenaline) epinephrine or other sympathomimetic amines that may precipitate ventricular arrhythmias. Tachyarrhythmias caused by increased myocardial sensitisation may be treated with propranolol, 1-2 mg IV or esmolol 25-100 microgm/kg/min IV.
- Monitor the ECG for 4-6 hours
- B: Specific drugs and antidotes:
- There is no specific antidote
- C: Decontamination
- Inhalation; remove victim from exposure, and give supplemental oxygen if available.
- Ingestion; (a) Prehospital: Administer activated charcoal, if available. DO NOT induce vomiting because of rapid absorption and the risk of abrupt onset CNS depression. (b) Hospital: Administer activated charcoal, although the efficacy of charcoal is unknown. Perform gastric lavage only if the ingestion was very large and recent (less than 30 minutes) D: Enhanced elimination:
- There is no documented efficacy for diuresis, haemodialysis, haemoperfusion, or repeat-dose charcoal
- POISONING and DRUG OVERDOSE, Californian Poison Control System Ed. Kent R Olson; 3rd Edition
- Do not administer sympathomimetic drugs unless absolutely necessary as material may increase myocardial irritability.
- No specific antidote.
- Because rapid absorption may occur through lungs if aspirated and cause systematic effects, the decision of whether to induce vomiting or not should be made by an attending physician.
- If lavage is performed, suggest endotracheal and/or esophageal control.
- Danger from lung aspiration must be weighed against toxicity when considering emptying the stomach.
- Treatment based on judgment of the physician in response to reactions of the patient
- To treat poisoning by the higher aliphatic alcohols (up to C7):
- Gastric lavage with copious amounts of water.
- It may be beneficial to instill 60 ml of mineral oil into the stomach.
- Oxygen and artificial respiration as needed.
- Electrolyte balance: it may be useful to start 500 ml. M/6 sodium bicarbonate intravenously but maintain a cautious and conservative attitude toward electrolyte replacement unless shock or severe acidosis threatens.
- To protect the liver, maintain carbohydrate intake by intravenous infusions of glucose.
- + Haemodialysis if coma is deep and persistent. [GOSSELIN, SMITH HODGE: Clinical Toxicology of Commercial Products, Ed 5)

BASIC TREATMENT

- Establish a patent airway with suction where necessary.
- Watch for signs of respiratory insufficiency and assist ventilation as necessary.
- Administer oxygen by non-rebreather mask at 10 to 15 l/min.
- Monitor and treat, where necessary, for shock.
- Monitor and treat, where necessary, for pulmonary oedema.
- Anticipate and treat, where necessary, for seizures.
- DO NOT use emetics. Where ingestion is suspected rinse mouth and give up to 200 ml water (5 ml/kg recommended) for dilution where patient is able to swallow, has a strong gag reflex and does not drool.
- Give activated charcoal.

ADVANCED TREATMENT

- Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
- Positive-pressure ventilation using a bag-valve mask might be of use
- Monitor and treat, where necessary, for arrhythmias.
- Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications.
- If the patient is hypoglycaemic (decreased or loss of consciousness, tachycardia, pallor, dilated pupils, diaphoresis and/or dextrose strip or glucometer readings below 50 mg), give 50% dextrose.
- Hypotension with signs of hypovolaemia requires the cautious administration of fluids. Fluid overload might create complications.
- Drug therapy should be considered for pulmonary oedema.
- Treat seizures with diazepam.
- Proparacaine hydrochloride should be used to assist eye irrigation.

EMERGENCY DEPARTMENT

Laboratory analysis of complete blood count, serum electrolytes, BUN, creatinine, glucose, urinalysis, baseline for serum aminotransferases (ALT and AST), calcium, phosphorus and magnesium, may assist in establishing a treatment regime. Other useful analyses include anion and osmolar gaps, arterial blood gases (ABGs), chest radiographs and

electrocardiograph

- Positive end-expiratory pressure (PEEP)-assisted ventilation may be required for acute parenchymal injury or adult respiratory distress syndrome.
- Acidosis may respond to hyperventilation and bicarbonate therapy.
- Haemodialysis might be considered in patients with severe intoxication.
 Consult a toxicologist as necessary. BRONSTEIN, A.C. and CURRANCE, P.L. EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

For C8 alcohols and above.

Symptomatic and supportive therapy is advised in managing patients. For gas exposures:

BASIC TREATMENT

- Establish a patent airway with suction where necessary.
- Watch for signs of respiratory insufficiency and assist ventilation as necessary.
- Administer oxygen by non-rebreather mask at 10 to 15 l/min.
- Monitor and treat, where necessary, for pulmonary oedema .
- Monitor and treat, where necessary, for shock.
- Anticipate seizures.

ADVANCED TREATMENT

Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.

- Positive-pressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.
- Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications.
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- Treat seizures with diazepam.
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BRONSTEIN, A.C. and CURRANCE, P.L.

EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

SECTION 5 Fire-fighting measures

Extinguishing media

- Alcohol stable foam.
- Dry chemical powder.
- BCF (where regulations permit).

Special hazards arising from the substrate or mixture

Fire Incompatibility + Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

Special protective equipment and precautions for fire-fighters

Fire Fighting	GENERAL
	 Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Fight fire from a safe distance, with adequate cover.
	 Containers may explode when heated - Ruptured cylinders may rocket May burn but does not ignite easily. Fire exposed cylinders may vent contents through pressure relief devices thereby increasing vapour concentration Fire may produce irritating, poisonous or corrosive gases. Decomposition may produce toxic fumes of:
Fire/Explosion Hazard	, carbon monoxide (CO) , carbon dioxide (CO2) , hydrogen fluoride , other pyrolysis products typical of burning organic material.

SECTION 6 Accidental release measures

Personal precautions, protective equipment and emergency procedures

See section 8

Environmental precautions

See section 12

Methods and material for containment and cleaning up

Minor Spills	 Avoid breathing vapour and any contact with liquid or gas. Protective equipment including respirator should be used. DO NOT enter confined spaces where gas may have accumulated.
Major Spills	 Clear area of all unprotected personnel and move upwind. Alert Emergency Authority and advise them of the location and nature of hazard. Wear full body clothing with breathing apparatus. Remove leaking cylinders to a safe place. Fit vent pipes. Release pressure under safe, controlled conditions Burn issuing gas at vent pipes. DO NOT exert excessive pressure on valve; DO NOT attempt to operate damaged valve.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

SECTION 7 Handling and storage

recautions for safe handling Safe handling	Consider use in closed pressurised systems, fitted with temperature, pressure and safety relief valves which are vented for safe dispersal. Use only properly specified equipment which is suitable for this product, its supply pressure and temperature The tubing network design connecting gas cylinders to the delivery system should include appropriate pressure indicators and vacuum or suction lines. Fully-welded types of pressure gauges, where the bourdon tube sensing element is welded to the gauge body, are recommended.	
Other information	 DO NOT transfer gas from one cylinder to another. Cylinders should be stored in a purpose-built compound with good ventilation, preferably in the open. Such compounds should be sited and built in accordance with statutory requirements. The storage compound should be kept clear and access restricted to authorised personnel only. 	

Conditions for safe storage, including any incompatibilities

Suitable container	 Cylinder: Ensure the use of equipment rated for cylinder pressure. Ensure the use of compatible materials of construction. Valve protection cap to be in place until cylinder is secured, connected.
Storage incompatibility	 Glycols and their ethers undergo violent decomposition in contact with 70% perchloric acid. This seems likely to involve formation of the glycol perchlorate esters (after scission of ethers) which are explosive, those of ethylene glycol and 3-chloro-1,2-propanediol being more powerful than glyceryl nitrate, and the former so sensitive that it explodes on addition of water. As a general rule, hydrofluorocarbons tend to be flammable unless they contain more fluorine atoms than hydrogen atoms. Alcohols are incompatible with strong acids, acid chlorides, acid anhydrides, oxidising and reducing agents. reacts, possibly violently, with alkaline metals and alkaline earth metals to produce hydrogen react with strong acids, strong caustics, aliphatic amines, isocyanates, acetaldehyde, benzoyl peroxide, chromic acid, chromium oxide, dialkylzincs, dichlorine oxide, ethylene oxide, hypochlorous acid, isopropyl chlorocarbonate, lithium tetrahydroaluminate, nitrogen dioxide, pentafluoroguanidine, phosphorus halides, phosphorus pentasulfide, tangerine oil, triethylaluminium, triisobutylaluminium should not be heated above 49 deg. C. when in contact with aluminium equipment Avoid magnesium, aluminium and their alloys, brass and steel. Compressed gases may contain a large amount of kinetic energy over and above that potentially available from the energy of reaction produced by the gas in chemical reaction with other substances

SECTION 8 Exposure controls / personal protection

Control parameters

Occupational Exposure Limits (OEL)

INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Limits (PELs) Table Z-1	dibutyltin dilauryl mercaptide	Tin, organic compounds (as Sn)	0.1 mg/m3	Not Available	Not Available	Not Available
US NIOSH Recommended Exposure Limits (RELs)	dibutyltin dilauryl mercaptide	Tin (organic compounds, as Sn)	0.1 mg/m3	Not Available	Not Available	[skin] [*Note: The REL applies to all organic tin compounds except Cyhexatin.]
US NIOSH Recommended Exposure Limits (RELs)	ethylene glycol	Ethylene glycol	Not Available	Not Available	Not Available	See Appendix D

Emergency Limits

Ingredient	TEEL-1	TEEL-2		TEEL-3
diethylene glycol	6.9 ppm	140 ppm		860 ppm
ethylene glycol	30 ppm	150 ppm		900 ppm
1,3,3,3-tetrafluoropropene	1,400 ppm	Not Available		Not Available
Ingredient	Original IDLH		Revised IDLH	
diethylene glycol	Not Available		Not Available	
tris(2-chloroisopropyl)phosphate	Not Available		Not Available	
dibutyltin dilauryl mercaptide	25 mg/m3		Not Available	
ethylene glycol	Not Available		Not Available	
N-methyldicyclohexylamine	Not Available		Not Available	
1,3,3,3-tetrafluoropropene	Not Available		Not Available	

Occupational Exposure Banding

Ingredient	Occupational Exposure Band Rating	Occupational Exposure Band Limit	
diethylene glycol	E	≤ 0.1 ppm	
tris(2-chloroisopropyl)phosphate	E	≤ 0.1 ppm	
Notes:	Occupational exposure banding is a process of assigning chemicals into specific categories or bands based on a chemical's potency and the		

adverse health outcomes associated with exposure. The output of this process is an occupational exposure band (OEB), which corresponds to a range of exposure concentrations that are expected to protect worker health.

Appropriate engineering controls	Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are: Process controls which involve changing the way a job activity or process is done to reduce the risk.		
Personal protection			
Eye and face protection	 Safety glasses with side shields. Chemical goggles. Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants. 		
Skin protection	See Hand protection below		
Hands/feet protection	 NOTE: The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact. Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed. When handling sealed and suitably insulated cylinders wear cloth or leather gloves. 		
Body protection	See Other protection below		
Other protection	 Protective overalls, closely fitted at neck and wrist. Eye-wash unit. Ensure availability of lifeline in confined spaces. 		

Respiratory protection

Full face respirator with supplied air.

- Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used Positive pressure, full face, air-supplied breathing apparatus should be used for work in enclosed spaces if a leak is suspected or the primary containment is to be opened (e.g. for
- a cylinder change)
- Air-supplied breathing apparatus is required where release of gas from primary containment is either suspected or demonstrated.

SECTION 9 Physical and chemical properties

Information on basic physical and chemical properties

Appearance	Not Available		
Physical state	Compressed Gas	Relative density (Water = 1)	Not Available
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	Not Available	Molecular weight (g/mol)	Not Available
Flash point (°C)	Not Available	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Available	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Immiscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

SECTION 10 Stability and reactivity

Reactivity	See section 7
Chemical stability	 Unstable in the presence of incompatible materials. Product is considered stable. Hazardous polymerisation will not occur.
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7

Hazardous decomposition products

SECTION 11 Toxicological information

See section 5

Information on toxicological effects

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Inhaled	The material is not thought to produce respiratory irritation (as classified by EC Directives using animal models). Nevertheless inhalation of the material, especially for prolonged periods, may produce respiratory discomfort and occasionally, distress. Inhalation of vapours or aerosols (mists, fumes), generated by the material during the course of normal handling, may be damaging to the health of the individual. Exposure to fluorocarbons can produce non-specific flu-like symptoms such as chills, fever, weakness, muscle pain, headache, chest discomfort, sore throat and dry cough with rapid recovery. High concentrations can cause irregular heartbeats and a stepwise reduction in lung capacity. Inhalation of the vapour is hazardous and may even be fatal The use of a quantity of material in an unventilated or confined space may result in increased exposure and an irritating atmosphere developing. Before starting consider control of exposure by mechanical ventilation. Aliphatic alcohols with more than 3-carbons cause headache, dizziness, drowsiness, muscle weakness and delirium, central depression, coma, seizures and behavioural changes. Secondary respiratory depression and failure, as well as low blood pressure and irregular heart rhythms, may follow. Inhalation of toxic gases may cause: Central Nervous System effects including depression, headache, confusion, dizziness, stupor, coma and seizures; respiratory: acute lung swellings, shortness of breath, wheezing, rapid breathing, other symptoms and respiratory arrest; heart: collapse, irregular heartbeats and cardiac arrest; gastrointestinal: irritation, ulcers, nausea and vomiting (may be bloody), and abdominal pain.
Ingestion	Accidental ingestion of the material may be harmful; animal experiments indicate that ingestion of less than 150 gram may be fatal or may produce serious damage to the health of the individual. Overexposure to non-ring alcohols causes nervous system symptoms. These include headache, muscle weakness and inco-ordination, giddiness, confusion, delirium and coma. Not normally a hazard due to physical form of product. Considered an unlikely route of entry in commercial/industrial environments
Skin Contact	This material can cause inflammation of the skin on contact in some persons. The material may accentuate any pre-existing dermatitis condition Fluorocarbons remove natural oils from the skin, causing irritation, dryness and sensitivity. Most liquid alcohols appear to act as primary skin irritants in humans. Significant percutaneous absorption occurs in rabbits but not apparently in man. Open cuts, abraded or irritated skin should not be exposed to this material Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.
Eye	This material can cause eye irritation and damage in some persons. Not considered to be a risk because of the extreme volatility of the gas.
Chronic	Repeated or long-term occupational exposure is likely to produce cumulative health effects involving organs or biochemical systems. Skin contact with the material is more likely to cause a sensitisation reaction in some persons compared to the general population. Ample evidence exists from experimentation that reduced human fertility is directly caused by exposure to the material. The reactivity of an epoxide intermediate may be the reason for the cancer-causing properties of halogenated oxiranes. It is reported that 1,1-dichloroethyne, vinyl chloride, trichloroethylene, tetrachloroethylene and chloroprene all cause cancer. Generally speaking, substances with one halogen substitution show higher potential to cause cancer compared to substances with two. Main route of exposure to the gas in the workplace is by inhalation. There has been some concern that this material can cause cancer or mutations but there is not enough data to make an assessment. Fluorocarbons can cause an increased risk of cancer, spontaneous abortion and birth defects.

HandiFoam HFO FR Black	ΤΟΧΙCΙΤΥ	IRRITATION
B-side	Not Available	Not Available
	ΤΟΧΙΟΙΤΥ	IRRITATION
	Dermal (rabbit) LD50: 11890 mg/kg ^[2]	Eye (rabbit) 50 mg mild
	Inhalation(Rat) LC50: >4.6 mg/l4h ^[1]	Eye: no adverse effect observed (not irritating) ^[1]
diethylene glycol	Oral (Rat) LD50; 12565 mg/kg ^[2]	Skin (human): 112 mg/3d-l mild
		Skin (rabbit): 500 mg mild
		Skin: no adverse effect observed (not irritating) ^[1]
	ΤΟΧΙΟΙΤΥ	IRRITATION
	Dermal (rabbit) LD50: >5000 mg/kg* ^[2]	Eye (rabbit): non-irritating*
tris(2- chloroisopropyl)phosphate	Inhalation(Rat) LC50: >4.6 mg/kl/4H* ^[2]	Skin (rabbit): mild (24 h):
chlorolsopropyrphiosphate	Intravenous (Mouse) LD50: 56 mg/kg ^[2]	
	Oral (Rat) LD50; 1500 mg/kg ^[2]	
	ΤΟΧΙΟΙΤΥ	IRRITATION
dibutyltin dilauryl mercaptide	Dermal (rabbit) LD50: >1000<2000 mg/kg ^[1]	Eye: no adverse effect observed (not irritating) ^[1]
	Oral (Rat) LD50; 136 mg/kg ^[2]	Skin: adverse effect observed (irritating) ^[1]
	ΤΟΧΙΟΙΤΥ	IRRITATION
	dermal (mouse) LD50: >3500 mg/kg ^[1]	Eye (rabbit): 100 mg/1h - mild
ethylene glycol	Oral (Rat) LD50; >2000 mg/kg ^[2]	Eye (rabbit): 12 mg/m3/3D
		Eye (rabbit): 1440mg/6h-moderate

		Eye (rabbit): 500 mg/24h - mild			
		Eye: no adverse effect observed (not irritating) ^[1]			
		Skin (rabbit): 555 mg(open)-mild			
		Skin: no adverse effect observed (not irritating) ^[1]			
	ΤΟΧΙΟΙΤΥ	IRRITATION			
	Dermal (rabbit) LD50: 323 mg/kg ^[2]	Not Available			
N-methyldicyclohexylamine	Inhalation(Rat) LC50: >0.54 mg/L4h ^[2]				
	Oral (Rat) LD50; >=267 mg/kg ^[1]				
	ΤΟΧΙΟΙΤΥ	IRRITATION			
1,3,3,3-tetrafluoropropene	Inhalation(Rat) LC50: >1157.752 ppm4h ^[2]	Not Available			
Legend:	1. Value obtained from Europe ECHA Registered Substances - A specified data extracted from RTECS - Register of Toxic Effect o	Acute toxicity 2. Value obtained from manufacturer's SDS. Unless otherwise f chemical Substances			
HandiFoam HFO FR Black B-side	The following information refers to contact allergens as a gro Contact allergies quickly manifest themselves as contact ecz eczema involves a cell-mediated (T lymphocytes) immune re	ema, more rarely as urticaria or Quincke's oedema. The pathogenesis of contact			
DIETHYLENE GLYCOL	Diglycolic acid is formed following the oxidation of accidental with fatal outcome.	y ingested diethylene glycol in the body and can lead to severe complications			
tris(2-chloroisopropyl)phosphate	source of potential exposure (human and environmental) to t ingredient in rubber or plastic to the outer surface after curing For tris(2-chloro-1-methylethyl)phosphate (TCPP) The flame retardant product supplied in the EU, marketed as isomers in this reaction mixture are not separated or markete Alkyl esters of phosphoric acid exhibit a low to moderate acu	The flame retardant product supplied in the EU, marketed as TCPP, is actually a reaction mixture containing four isomers. The individual isomers in this reaction mixture are not separated or marketed. The individual components are never produced as such. Alkyl esters of phosphoric acid exhibit a low to moderate acute toxicity and metabolised. From studies done on mice, they are not likely to cause gene damage or affect reproduction. However, 2-ethylhexanoic acid produced an effect on newborn rats at high doses to the pregnant			
DIBUTYLTIN DILAURYL MERCAPTIDE	* Arkema Canada MSDS for substance in 40% mineral oil **	Geleste MSDS			
ETHYLENE GLYCOL	[Estimated Lethal Dose (human) 100 ml; RTECS quoted by Orica] Substance is reproductive effector in rats (birth defects). Mutagenic to rat cells. For ethylene glycol: Ethylene glycol is quickly and extensively absorbed throughout the gastrointestinal tract. Limited information suggests that it is also absorbed through the airways; absorption through skin is apparently slow. Following absorption, it is distributed throughout the body.				
N-METHYLDICYCLOHEXYLAMINE	Overexposure to most of these materials may cause adverse health effects. Many amine-based compounds can cause release of histamines, which, in turn, can trigger allergic and other physiological effects, including constriction of the bronchi or asthma and inflammation of the cavity of the nose. Whole-body symptoms include headache, nausea, faintness, anxiety, a decrease in blood pressure, rapid heartbeat, itching, reddening of the skin, urticaria (hives) and swelling of the face, which are usually transient. There are generally four routes of possible or potential exposure: inhalation, skin contact, eye contact, and swallowing. Inhalation: Inhaling vapours may result in moderate to severe irritation of the tissues of the nose and throat and can irritate the lungs. Higher concentrations of certain amines can produce severe respiratory irritation, characterized by discharge from the nose, coughing, difficulty in breathing and chest pain. The material may be irritating to the eye, with prolonged contact causing inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis. Asthma-like symptoms may continue for months or even years after exposure to the material ends. This may be due to a non-allergic condition known as reactive airways dysfunction syndrome (RADS) which can occur after exposure to high levels of highly irritating compound. Main criteria RDB include the absence of previous airways disease in a non-atopic individual, with sudden onset of persistent asthma-like symptoms recorded. When applied to the skin of male rabbits, most adverse effects were observed within an hour after treatment and lasted several hours. The onset of paralysis occurred between several hours and two days after identification of the slightly irritating and the non-infritating test anticle concentrations in the primary irritation experiments, a main study was performed with the selected test substance while the control animals were intrader mally injected with a 5 % concentration. The epidermal exposure, the				
1,3,3,3-TETRAFLUOROPROPENE	Inhalation (rat) NOEL (28 days): >1.5 mg/l * Vendor HFO-1234ze is not likely to accumulate in the bodies of humans or animals HFO-1234ze is practically non-toxic. Short-term exposures at levels higher than 10% have not induced cardiac sensitization to adrenalin nor induced serious toxic effects. Rats and rabbits did not exhibit any serious toxic, developmental or reproductive effects even with exposures to high levels of HFO-1234ze. Based on a series of mutagenicity and genomics studies, the cancer risk for HFO-1234ze is low, no cardiac sensitisation was observed in dogs with exposures up to 120,000 ppm; repeated dose toxicity in rats (13-wk) found mild effects on the heart (NOEL 5,000ppm); in vitro genotoxicity findings include negative Ames Test and negative human lymphocyte chromosome aberration test; in vivo genotoxicity findings in the mouse micronucleus test were negative (inhalation, mammalian bone-marrow cytogenic test with				

	chromosomal analysis).				
HandiFoam HFO FR Black B-sic & 1,3,3, TETRAFLUOROPROPEN	 pressure and the production of blood cells. The p Disinfection byproducts (DBPs) are formed when matter in water. Animal studies have shown that s 	alation of perfluoroalkenes can cause lung injury, kidney damage, brain changes and death. Repeated exposures may alter blood ssure and the production of blood cells. The potential for causing cancer is the subject of speculation. infection byproducts (DBPs) are formed when disinfectants such as chlorine, chloramines and ozone react with organic and inorganic tter in water. Animal studies have shown that some DBPs cause cancer. To date, several hundred DBPs have been identified. merous haloalkanes and haloalkenes have been tested for cancer-causing and mutation-causing activities.			
DIETHYLENE GLYCOL N-METHYLDICYCLOHEXYLAMIN		The material may cause skin irritation after prolonged or repeated exposure and may produce on contact skin redness, swelling, the production of vesicles, scaling and thickening of the skin.			
Acute Toxicity	×	Carcinogenicity	×		
Skin Irritation/Corrosion	✓	Reproductivity	✓		
Serious Eye Damage/Irritation	✓	STOT - Single Exposure	×		
Respiratory or Skin sensitisation	✓	STOT - Repeated Exposure	×		
Mutagenicity	×	Aspiration Hazard	×		
		Legend: 🗙 – Data either r	not available or does not fill the criteria for classification		

Data Gallable to make classification

SECTION 12 Ecological information

Toxicity

HandiFoam HFO FR Black	Endpoint	Test Duration (hr)	Species		Value	Source
B-side	Not Available	Not Available	Not Available		Not Available	Not Availabl
	Endpoint	Test Duration (hr)	Species	Val	ue	Sourc
	EC50	48h	Crustacea	840	00mg/l	1
diethylene glycol	NOEC(ECx)	192h	Algae or other aquatic plants	800)mg/l	1
	LC50	96h	Fish	>10	0mg/l	4
	EC50	96h	Algae or other aquatic plants	650	00-13000mg/l	2
	Endpoint	Test Duration (hr)	Species		Value	Source
	BCF	1008h	Fish		0.8-2.8	7
	EC50	72h	Algae or other aquatic plants		82mg/l	Not Availab
tris(2-	ErC50	72h	Algae or other aquatic plants		4mg/l	1
chloroisopropyl)phosphate	EC50	48h	Crustacea		65335mg/l	1
	EC50(ECx)	96h	Algae or other aquatic plants		4mg/l	1
	LC50	96h	Fish		56.2mg/l	Not Availab
	EC50	96h	Algae or other aquatic plants		4mg/l	1
	Endpoint	Test Duration (hr)	Species	Species		Sourc
	EC50	72h	Algae or other aquatic plants		>=1.6mg/l	2
butyltin dilauryl mercaptide	EC50	48h	Crustacea	Crustacea		2
	EC50(ECx)	48h	Crustacea	Crustacea		2
	Endpoint	Test Duration (hr)	Species	Val	ue	Sourc
	EC50(ECx)	Not Available	Algae or other aquatic plants	650	0-7500mg/l	1
ethylene glycol	EC50	48h	Crustacea	>10	0mg/l	2
	LC50	96h	Fish	>10)000mg/l	1
	EC50	96h	Algae or other aquatic plants	650	0-13000mg/l	1
	Endpoint	Test Duration (hr)	Species		Value	Source
	EC50(ECx)	72h	Algae or other aquatic plants		0.063mg/l	Not Availab
N-methyldicyclohexylamine	EC50	72h	Algae or other aquatic plants	Algae or other aquatic plants 0.063mg		Not Availab
	EC50	48h	Crustacea	Crustacea 8mg/l		Not Availab
	LC50	96h	Fish		62mg/l	Not Availab
	Endpoint	Test Duration (hr)	Species	Species Valu		Sourc
1,3,3,3-tetrafluoropropene	EC50	72h	Algae or other aquatic plants		>170mg/l	2
1,3,3,3-tetranuoropropene						2

	EC50(ECx)	48h		Crustacea		>160mg/l	2
Legend:	Ecotox databas	1. IUCLID Toxicity Data 2. Eu e - Aquatic Toxicity Data 5. E ion Data 8. Vendor Data	, 0		5	,	,

In addition to carbon dioxide (CO2), methane (CH4) and nitrous oxide (N2O), the greenhouse gases mentioned in the Kyoto Protocol include synthetic substances that share the common feature of being highly persistent in the atmosphere and inhibit radiation from escaping out of the atmosphere. These synthetic substances include hydrocarbons that are partially fluorinated (HCFs) or totally fluorinated (PFCs) as well as sulfur hexafluoride (SF6). The greenhouse potential of these substances, expressed as multiples of that of CO2, are within the range of 140 to 11,700 for HFCs, from 6500 to 9,200 for PFCs and 23,900 for SF6. **DO NOT** discharge into sewer or waterways.

Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
diethylene glycol	LOW	LOW
tris(2-chloroisopropyl)phosphate	HIGH	HIGH
dibutyltin dilauryl mercaptide	HIGH	HIGH
ethylene glycol	LOW (Half-life = 24 days)	LOW (Half-life = 3.46 days)
N-methyldicyclohexylamine	HIGH	HIGH

Bioaccumulative potential

Ingredient	Bioaccumulation
diethylene glycol	LOW (BCF = 180)
tris(2-chloroisopropyl)phosphate	LOW (BCF = 4.6)
dibutyltin dilauryl mercaptide	LOW (LogKOW = 16.42)
ethylene glycol	LOW (BCF = 200)
N-methyldicyclohexylamine	LOW (LogKOW = 3.71)

Mobility in soil

Ingredient	Mobility
diethylene glycol	HIGH (KOC = 1)
tris(2-chloroisopropyl)phosphate	LOW (KOC = 1278)
dibutyltin dilauryl mercaptide	LOW (KOC = 6187000000)
ethylene glycol	HIGH (KOC = 1)
N-methyldicyclohexylamine	LOW (KOC = 325)

SECTION 13 Disposal considerations

Waste treatment methods	
Product / Packaging disposal	 Evaporate residue at an approved site. Return empty containers to supplier. If containers are marked non-returnable establish means of disposal with manufacturer prior to purchase.

SECTION 14 Transport information

Labels Required



Marine Pollutant

Land transport (DOT)

Land transport (DOT)			
UN number	3500		
UN proper shipping name	nemical under pressure, n.o.s.(Contains Hydrofluoroolefin, Nitrogen)		
Transport hazard class(es)	Class 2.2 Subrisk Not Applicable		
Packing group	Not Applicable		
Environmental hazard	Not Applicable		
Special precautions for user	Hazard Label2.2Special provisions362, T50, TP40		

UN number	3500			
UN proper shipping name	Chemical under pressur	e, n.o.s.(Contains Hydrofluoroolefin, Nit	ogen)	
	ICAO/IATA Class	2.2		
Transport hazard class(es)	ICAO / IATA Subrisk	Not Applicable		
	ERG Code	2L		
Packing group	Not Applicable			
Environmental hazard	Not Applicable			
	Special provisions		A187	
	Cargo Only Packing Instructions		218	
	Cargo Only Maximum Qty / Pack		150 kg	
Special precautions for user	Passenger and Cargo	Passenger and Cargo Packing Instructions		
	Passenger and Cargo Maximum Qty / Pack		75 kg	
	Passenger and Cargo	Limited Quantity Packing Instructions	Forbidden	

Sea transport (IMDG-Code / GGVSee)

UN number	500	
UN proper shipping name	CHEMICAL UNDER PRESSURE, N.O.S.(Contair	s Hydrofluoroolefin, Nitrogen)
Transport hazard class(es)	IMDG Class 2.2 IMDG Subrisk Not Applicable	
Packing group	Not Applicable	
Environmental hazard	Not Applicable	
Special precautions for user	EMS NumberF-C, S-VSpecial provisions274 362Limited Quantities0	

Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
diethylene glycol	Not Available
tris(2-chloroisopropyl)phosphate	Not Available
dibutyltin dilauryl mercaptide	Not Available
ethylene glycol	Not Available
N-methyldicyclohexylamine	Not Available
1,3,3,3-tetrafluoropropene	Not Available

Transport in bulk in accordance with the ICG Code

Product name	Ship Type
diethylene glycol	Not Available
tris(2-chloroisopropyl)phosphate	Not Available
dibutyltin dilauryl mercaptide	Not Available
ethylene glycol	Not Available
N-methyldicyclohexylamine	Not Available
1,3,3,3-tetrafluoropropene	Not Available

SECTION 15 Regulatory information

Safety, health and environmental regulations / legislation specific for the substance or mixture

diethylene glycol is found on the following regulatory lists

- US AIHA Workplace Environmental Exposure Levels (WEELs)
- US DOE Temporary Emergency Exposure Limits (TEELs)
- US Toxic Substances Control Act (TSCA) Chemical Substance Inventory

tris(2-chloroisopropyl)phosphate is found on the following regulatory lists

US - California - Biomonitoring - Priority Chemicals

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US Toxicology Excellence for Risk Assessment (TERA) Workplace Environmental Exposure Levels (WEEL)

US TSCA Chemical Substance Inventory - Interim List of Active Substances

US TSCA Chemical Substance Inventory - Interim List of Active Substances

dibutyltin dilauryl mercaptide is found on the following regulatory lists

Chemical Footprint Project - Chemicals of High Concern List US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory US TSCA Chemical Substance Inventory - Interim List of Active Substances US NIOSH Recommended Exposure Limits (RELs) US OSHA Permissible Exposure Limits (PELs) Table Z-1 ethylene glycol is found on the following regulatory lists Chemical Footprint Project - Chemicals of High Concern List US Clean Air Act - Hazardous Air Pollutants US - California Hazardous Air Pollutants Identified as Toxic Air Contaminants US DOE Temporary Emergency Exposure Limits (TEELs) US - California Proposition 65 - Maximum Allowable Dose Levels (MADLs) for US EPA Integrated Risk Information System (IRIS) Chemicals Causing Reproductive Toxicity` US EPCRA Section 313 Chemical List US - California Proposition 65 - Reproductive Toxicity US NIOSH Recommended Exposure Limits (RELs) US - California Safe Drinking Water and Toxic Enforcement Act of 1986 - Proposition 65 US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory List US TSCA Chemical Substance Inventory - Interim List of Active Substances US - Massachusetts - Right To Know Listed Chemicals US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs) N-methyldicyclohexylamine is found on the following regulatory lists US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory US TSCA Chemical Substance Inventory - Interim List of Active Substances 1,3,3,3-tetrafluoropropene is found on the following regulatory lists US AIHA Workplace Environmental Exposure Levels (WEELs) US Toxicology Excellence for Risk Assessment (TERA) Workplace Environmental US DOE Temporary Emergency Exposure Limits (TEELs) Exposure Levels (WEEL) US TSCA Chemical Substance Inventory - Interim List of Active Substances

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

Federal Regulations

Superfund Amendments and Reauthorization Act of 1986 (SARA)

Section 311/312 hazard categories

Flammable (Gases, Aerosols, Liquids, or Solids)	No
Gas under pressure	Yes
Explosive	No
Self-heating	No
Pyrophoric (Liquid or Solid)	No
Pyrophoric Gas	No
Corrosive to metal	No
Oxidizer (Liquid, Solid or Gas)	No
Organic Peroxide	No
Self-reactive	No
In contact with water emits flammable gas	No
Combustible Dust	No
Carcinogenicity	No
Acute toxicity (any route of exposure)	No
Reproductive toxicity	Yes
Skin Corrosion or Irritation	Yes
Respiratory or Skin Sensitization	Yes
Serious eye damage or eye irritation	Yes
Specific target organ toxicity (single or repeated exposure)	No
Aspiration Hazard	No
Germ cell mutagenicity	No
Simple Asphyxiant	No
Hazards Not Otherwise Classified	No

Reportable Quantity in Pounds (Ib) Reportable Quantity in kg Name 5000 2270 ethylene glycol

State Regulations

US. California Proposition 65

WARNING: This product can expose you to chemicals including ethylene glycol, which is known to the State of California to cause birth defects or other reproductive harm. For more information, go to www.P65Warnings.ca.gov.

National Inventory Status

National Inventory	Status
Australia - AIIC / Australia Non-Industrial Use	Yes
Canada - DSL	Yes
Canada - NDSL	No
China - IECSC	Yes

National Inventory	Status	
Europe - EINEC / ELINCS / NLP	Yes	
Japan - ENCS	Yes	
Korea - KECI	No (N-methyldicyclohexylamine)	
New Zealand - NZIoC	No (1,3,3,3-tetrafluoropropene)	
Philippines - PICCS	No (1,3,3,3-tetrafluoropropene)	
USA - TSCA	Yes	
Taiwan - TCSI	Yes	
Mexico - INSQ	No (N-methyldicyclohexylamine; 1,3,3,3-tetrafluoropropene)	
Vietnam - NCI	Yes	
Russia - FBEPH	Yes	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.	

SECTION 16 Other information

Revision Date	11/09/2022
Initial Date	01/28/2022

CONTACT POINT

PLEASE NOTE THAT TITANIUM DIOXIDE IS NOT PRESENT IN CLEAR OR NEUTRAL BASES

SDS Version Summary

Version	Date of Update	Sections Updated
2.6	11/09/2022	Ingredients, Physical Properties

Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chernwatch Classification committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios.

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