

# HandiFoam FR HFO & HandiFoam FR Black HFO A-side ICP Construction Inc.

Version No: 2.4

Safety Data Sheet according to OSHA HazCom Standard (2012) requirements

Issue Date: 11/09/2022 Print Date: 11/09/2022 S.GHS.USA.EN

### **SECTION 1 Identification**

### **Product Identifier**

Product name	HandiFoam FR HFO & HandiFoam FR Black HFO A-side			
Synonyms	ot Available			
Proper shipping name	Chemical under pressure, n.o.s. (Hydrofluoroolefin, Nitrogen)			
Other means of identification	entification Not Available			

#### Recommended use of the chemical and restrictions on use

Relevant identified uses Low pressure polyurethane foam. Side A Component. For PROFESSIONAL USE ONLY

### Name, address, and telephone number of the chemical manufacturer, importer, or other responsible party

Registered company name	ICP Construction Inc.			
Address	50 Dascomb Road Andover, MA 01810 United States			
Telephone	667-5119 1-978-623-9987			
Fax	Not Available			
Website	www.icpgroup.com			
Email	sds@icpgroup.com			

### Emergency phone number

• • •				
Association / Organisation	ChemTel			
Emergency telephone numbers	1-800-255-3924			
Other emergency telephone numbers	1-813-248-0585			

### SECTION 2 Hazard(s) identification

### Classification of the substance or mixture



Note: The hazard category numbers found in GHS classification in section 2 of this SDSs are NOT to be used to fill in the NFPA 704 diamond. Blue = Health Red = Fire Yellow = Reactivity White = Special (Oxidizer or water reactive substances)

Classification

Serious Eye Damage/Eye Irritation Category 2A, Sensitisation (Respiratory) Category 1, Specific Target Organ Toxicity - Repeated Exposure Category 2, Acute Toxicity (Inhalation) Category 4, Specific Target Organ Toxicity - Single Exposure (Respiratory Tract Irritation) Category 3, Skin Corrosion/Irritation Category 2, Gases Under Pressure (Compressed Gas), Sensitisation (Skin) Category 1, Carcinogenicity Category 2

### Label elements

Hazard pictogram(s)







Signal word

Danger

### Hazard statement(s)

H31

Causes serious eye irritation.

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H334	May cause allergy or asthma symptoms or breathing difficulties if inhaled.				
H373	ay cause damage to organs through prolonged or repeated exposure. (Respiratory system) (Inhalation)				
H332	Harmful if inhaled.				
H335	May cause respiratory irritation.				
H315	Causes skin irritation.				
H280	Contains gas under pressure; may explode if heated.				
H317	May cause an allergic skin reaction.				
H351	Suspected of causing cancer.				

### Hazard(s) not otherwise classified

Not Applicable

### Precautionary statement(s) Prevention

P202 Do not handle until all safety precautions have been read and understood				
P260	o not breathe gas.			
P262	Do not get in eyes, on skin, or on clothing			
P264	Wash hands and other skin areas exposed to material thoroughly after handling			
P271	Use only outdoors or in a well-ventilated area.			
P280	Wear protective gloves, protective clothing and eye protection			
P284	Wear respiratory protection			

#### Precautionary statement(s) Response

P308+P313	IF exposed or concerned: Get medical advice/ attention.				
P342+P311	P342+P311 If experiencing respiratory symptoms: Call a POISON CENTER/doctor/physician/first aider.				
P305+P351+P338	i+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.				
P304+P340	P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.				
P302+P352+P333+P313 IF ON SKIN: Wash with plenty of soap and water. If skin irritation or rash occurs: Get medical attention					

### Precautionary statement(s) Storage

P405	Store locked up.	
P410+P403	Protect from sunlight. Store in a well-ventilated place.	
P403+P233 Store in a well-ventilated place. Keep container tightly closed.		

### Precautionary statement(s) Disposal

P501 Dispose of contents/container to authorised hazardous or special waste collection point in accordance with any local regulation.

### **SECTION 3 Composition / information on ingredients**

### Substances

See section below for composition of Mixtures

### Mixtures

CAS No	%[weight]	Name	
101-68-8	30-60	4.4'-diphenylmethane diisocyanate (MDI)	
9016-87-9	30-60	polymeric diphenylmethane diisocyanate	
29118-24-9	5-10	1.3.3.3-tetrafluoropropene	
7727-37-9.	<5	nitrogen	

The specific chemical identity and/or exact percentage (concentration) of composition has been withheld as a trade secret.

### **SECTION 4 First-aid measures**

### Description of first aid measures

**Eye Contact** 

- If product comes in contact with eyes remove the patient from gas source or contaminated area.
- ▶ Take the patient to the nearest eye wash, shower or other source of clean water.
- Open the eyelid(s) wide to allow the material to evaporate.
- Gently rinse the affected eye(s) with clean, cool water for at least 15 minutes. Have the patient lie or sit down and tilt the head back. Hold the eyelid(s) open and pour water slowly over the eyeball(s) at the inner corners, letting the water run out of the outer corners.
- The patient may be in great pain and wish to keep the eyes closed. It is important that the material is rinsed from the eyes to prevent further damage.
- Ensure that the patient looks up, and side to side as the eye is rinsed in order to better reach all parts of the eye(s)
  - ► Transport to hospital or doctor.
  - Even when no pain persists and vision is good, a doctor should examine the eye as delayed damage may occur.
  - If the patient cannot tolerate light, protect the eyes with a clean, loosely tied bandage.
  - Ensure verbal communication and physical contact with the patient.

DO NOT allow the patient to rub the eyes

**DO NOT** allow the patient to tightly shut the eyes

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DO NOT introduce oil or ointment into the eye(s) without medical advice DO NOT use hot or tepid water. If skin or hair contact occurs: Immediately flush body and clothes with large amounts of water, using safety shower if available. Skin Contact Quickly remove all contaminated clothing, including footwear Wash skin and hair with running water. Continue flushing with water until advised to stop by the Poisons Information Centre. Transport to hospital, or doctor. Following uptake by inhalation, move person to an area free from risk of further exposure. Oxygen or artificial respiration should be administered as needed. Asthmatic-type symptoms may develop and may be immediate or delayed up to several hours. Treatment is essentially symptomatic. A physician should be consulted. Following exposure to gas, remove the patient from the gas source or contaminated area. NOTE: Personal Protective Equipment (PPE), including positive pressure self-contained breathing apparatus may be required to assure the safety of the rescuer. Prostheses such as false teeth, which may block the airway, should be removed, where possible, prior to initiating first aid procedures. If the patient is not breathing spontaneously, administer rescue breathing. Inhalation If the patient does not have a pulse, administer CPR. If medical oxygen and appropriately trained personnel are available, administer 100% oxygen. Summon an emergency ambulance. If an ambulance is not available, contact a physician, hospital, or Poison Control Centre for further instruction. Keep the patient warm, comfortable and at rest while awaiting medical care. MONITOR THE BREATHING AND PULSE, CONTINUOUSLY Administer rescue breathing (preferably with a demand-valve resuscitator, bag-valve mask-device, or pocket mask as trained) or CPR if Not considered a normal route of entry. Ingestion

### Most important symptoms and effects, both acute and delayed

See Section 11

### Indication of any immediate medical attention and special treatment needed

For gas exposures:

BASIC TREATMENT

- Establish a patent airway with suction where necessary.
- Watch for signs of respiratory insufficiency and assist ventilation as necessary.
- Administer oxygen by non-rebreather mask at 10 to 15 l/min.
- Monitor and treat, where necessary, for pulmonary oedema.
- Monitor and treat, where necessary, for shock
- Anticipate seizures.

ADVANCED TREATMENT

- ▶ Consider orotracheal or nasotracheal intubation for airway control in unconscious patient or where respiratory arrest has occurred.
- Positive-pressure ventilation using a bag-valve mask might be of use.
- Monitor and treat, where necessary, for arrhythmias.
- Start an IV D5W TKO. If signs of hypovolaemia are present use lactated Ringers solution. Fluid overload might create complications.
- Drug therapy should be considered for pulmonary oedema
- Hypotension with signs of hypovolaemia requires the cautious administration of fluids. Fluid overload might create complications.
- Treat seizures with diazepam.
- ▶ Proparacaine hydrochloride should be used to assist eye irrigation.

BRONSTEIN, A.C. and CURRANCE, P.L.

EMERGENCY CARE FOR HAZARDOUS MATERIALS EXPOSURE: 2nd Ed. 1994

For sub-chronic and chronic exposures to isocyanates:

- This material may be a potent pulmonary sensitiser which causes bronchospasm even in patients without prior airway hyperreactivity.
- Clinical symptoms of exposure involve mucosal irritation of respiratory and gastrointestinal tracts.
- ▶ Conjunctival irritation, skin inflammation (erythema, pain vesiculation) and gastrointestinal disturbances occur soon after exposure.
- Pulmonary symptoms include cough, burning, substernal pain and dyspnoea.
- Some cross-sensitivity occurs between different isocyanates
- Noncardiogenic pulmonary oedema and bronchospasm are the most serious consequences of exposure. Markedly symptomatic patients should receive oxygen, ventilatory support and an intravenous line.
- Treatment for asthma includes inhaled sympathomimetics (epinephrine [adrenalin], terbutaline) and steroids.
- Activated charcoal (1 g/kg) and a cathartic (sorbitol, magnesium citrate) may be useful for ingestion
- Mydriatics, systemic analgesics and topical antibiotics (Sulamyd) may be used for corneal abrasions.
- There is no effective therapy for sensitised workers.

[Ellenhorn and Barceloux; Medical Toxicology]

NOTE: Isocyanates cause airway restriction in naive individuals with the degree of response dependant on the concentration and duration of exposure. They induce smooth muscle contraction which leads to bronchoconstrictive episodes. Acute changes in lung function, such as decreased FEV1, may not represent sensitivity. [Karol & Jin, Frontiers in Molecular Toxicology, pp 56-61, 1992]

Personnel who work with isocyanates, isocyanate prepolymers or polyisocyanates should have a pre-placement medical examination and periodic examinations thereafter, including a pulmonary function test. Anyone with a medical history of chronic respiratory disease, asthmatic or bronchial attacks, indications of allergic responses, recurrent eczema or sensitisation conditions of the skin should not handle or work with isocyanates. Anyone who develops chronic respiratory distress when working with isocyanates should be removed from exposure and examined by a physician. Further exposure must be avoided if a sensitivity to isocyanates or polyisocyanates has developed.

### **SECTION 5 Fire-fighting measures**

### **Extinguishing media**

- Small quantities of water in contact with hot liquid may react violently with generation of a large volume of rapidly expanding hot sticky semi-solid foam.
- Presents additional hazard when fire fighting in a confined space.
- Cooling with flooding quantities of water reduces this risk

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- ► Dry chemical powder.
- ► BCF (where regulations permit).
- Carbon dioxide.

### Special hazards arising from the substrate or mixture

Fire Incompatibility

Avoid contamination with oxidising agents i.e. nitrates, oxidising acids, chlorine bleaches, pool chlorine etc. as ignition may result

### Special protective equipment and precautions for fire-fighters

### **GENERAL** Fire Fighting ▶ Alert Fire Brigade and tell them location and nature of hazard. Wear full body protective clothing with breathing apparatus. Fight fire from a safe distance, with adequate cover. - Combustible. - Moderate fire hazard when exposed to heat or flame. - When heated to high temperatures decomposes rapidly generating vapour which pressures and may then rupture containers with release of flammable and highly toxic isocvanate vapour. ▶ Containers may explode when heated - Ruptured cylinders may rocket May burn but does not ignite easily. Fire exposed cylinders may vent contents through pressure relief devices thereby increasing vapour concentration.. Fire may produce irritating, poisonous or corrosive gases. Fire/Explosion Hazard Decomposition may produce toxic fumes of: carbon monoxide (CO) carbon dioxide (CO2) isocyanates hydrogen cyanide and minor amounts of nitrogen oxides (NOx) other pyrolysis products typical of burning organic material.

### **SECTION 6 Accidental release measures**

### Personal precautions, protective equipment and emergency procedures

See section 8

### **Environmental precautions**

See section 12

### Methods and material for containment and cleaning up

Minor Spills	<ul> <li>Avoid breathing vapour and any contact with liquid or gas. Protective equipment including respirator should be used.</li> <li>DO NOT enter confined spaces where gas may have accumulated.</li> </ul>
	For isocyanate spills of less than 40 litres (2 m2):
	• Evacuate area from everybody not dealing with the emergency, keep them upwind and prevent further access, remove ignition sources and, if inside building, ventilate area as well as possible.
	▶ Notify supervision and others as necessary.
	<ul> <li>Put on personal protective equipment (suitable respiratory protection, face and eye protection, protective suit, gloves and impermeable boots).</li> </ul>
	Avoid contamination with water, alkalies and detergent solutions.
	<ul> <li>Material reacts with water and generates gas, pressurises containers with even drum rupture resulting.</li> </ul>
Major Spills	▶ DO NOT reseal container if contamination is suspected.
	▶ DO NOT touch the spill material
	► Clear area of all unprotected personnel and move upwind.
	Alert Emergency Authority and advise them of the location and nature of hazard.
	▶ Wear full body clothing with breathing apparatus.
	Remove leaking cylinders to a safe place.
	Fit vent pipes. Release pressure under safe, controlled conditions
	Burn issuing gas at vent pipes.
	DO NOT exert excessive pressure on valve; DO NOT attempt to operate damaged valve.

Personal Protective Equipment advice is contained in Section 8 of the SDS.

### **SECTION 7 Handling and storage**

Precautions for safe handling						
Safe handling	Consider use in closed pressurised systems, fitted with temperature, pressure and safety relief valves which are vented for safe dispersal. Use only properly specified equipment which is suitable for this product, its supply pressure and temperature  The tubing network design connecting gas cylinders to the delivery system should include appropriate pressure indicators and vacuum or suction lines.  Fully-welded types of pressure gauges, where the bourdon tube sensing element is welded to the gauge body, are recommended.  DO NOT transfer gas from one cylinder to another.					
Other information	Consider storage under inert gas.  Consider storage under inert gas.  Cylinders should be stored in a purpose-built compound with good ventilation, preferably in the open.  Such compounds should be sited and built in accordance with statutory requirements.  The storage compound should be kept clear and access restricted to authorised personnel only.					

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### Conditions for safe storage, including any incompatibilities

### Suitable container

Storage incompatibility

- Cylinder:
- Ensure the use of equipment rated for cylinder pressure.
- Ensure the use of compatible materials of construction.
- Valve protection cap to be in place until cylinder is secured, connected.
- nucleophiles including alcohols, amines, and even water. Upon treatment with an alcohol, an isocyanate forms a urethane linkage. ▶ A range of exothermic decomposition energies for isocyanates is given as 20-30 kJ/mol.
- The relationship between energy of decomposition and processing hazards has been the subject of discussion; it is suggested that values of

· Avoid reaction with water, alcohols and detergent solutions. Isocyanates are electrophiles, and as such they are reactive toward a variety of

- energy released per unit of mass, rather than on a molar basis (J/g) be used in the assessment. For example, in 'open vessel processes' (with man-hole size openings, in an industrial setting), substances with exothermic decomposition energies below 500 J/g are unlikely to present a danger, whilst those in 'closed vessel processes' (opening is a safety valve or bursting disk)
- present some danger where the decomposition energy exceeds 150 J/g. Compressed gases may contain a large amount of kinetic energy over and above that potentially available from the energy of reaction produced by the gas in chemical reaction with other substances

### **SECTION 8 Exposure controls / personal protection**

#### **Control parameters**

#### Occupational Exposure Limits (OEL)

#### INGREDIENT DATA

Source	Ingredient	Material name	TWA	STEL	Peak	Notes
US OSHA Permissible Exposure Limits (PELs) Table Z-1	4,4'-diphenylmethane diisocyanate (MDI)	Methylene bisphenyl isocyanate (MDI)	Not Available	Not Available	0.02 ppm / 0.2 mg/m3	Not Available
US NIOSH Recommended Exposure Limits (RELs)	4,4'-diphenylmethane diisocyanate (MDI)	Methylene bisphenyl isocyanate	0.005 ppm / 0.05 mg/m3	Not Available	0.020 (10-minute) ppm / 0.2 (10-minute) mg/m3	Not Available

#### **Emergency Limits**

Ingredient	TEEL-1	TEEL-2	TEEL-3
4,4'-diphenylmethane diisocyanate (MDI)	0.45 mg/m3	Not Available	Not Available
4,4'-diphenylmethane diisocyanate (MDI)	29 mg/m3	40 mg/m3	240 mg/m3
polymeric diphenylmethane diisocyanate	0.15 mg/m3	3.6 mg/m3	22 mg/m3
1,3,3,3-tetrafluoropropene	1,400 ppm	Not Available	Not Available
nitrogen	7.96E+05 ppm	8.32E+05 ppm	8.69E+05 ppm

Ingredient	Original IDLH	Revised IDLH
4,4'-diphenylmethane diisocyanate (MDI)	75 mg/m3	Not Available
polymeric diphenylmethane diisocyanate	Not Available	Not Available
1,3,3,3-tetrafluoropropene	Not Available	Not Available
nitrogen	Not Available	Not Available

### Occupational Exposure Banding

Ingredient	Occupational Exposure Band Rating	Occupational Exposure Band Limit	
polymeric diphenylmethane diisocyanate	E	≤ 0.1 ppm	
Notes:	Occupational exposure banding is a process of assigning chemicals into specific categories or bands based on a chemical's potency and the		

### adverse health outcomes associated with exposure. The output of this process is an occupational exposure band (OEB), which corresponds to a range of exposure concentrations that are expected to protect worker health.

### **Exposure controls**

#### Appropriate engineering controls

Engineering controls are used to remove a hazard or place a barrier between the worker and the hazard. Well-designed engineering controls can be highly effective in protecting workers and will typically be independent of worker interactions to provide this high level of protection. The basic types of engineering controls are:

Process controls which involve changing the way a job activity or process is done to reduce the risk.

## Personal protection











- Eye and face protection
- Safety glasses with side shields.
- Chemical goggles.
- Contact lenses may pose a special hazard; soft contact lenses may absorb and concentrate irritants

### Skin protection

See Hand protection below

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Hands/feet protection	NOTE:  The material may produce skin sensitisation in predisposed individuals. Care must be taken, when removing gloves and other protective equipment, to avoid all possible skin contact.  Contaminated leather items, such as shoes, belts and watch-bands should be removed and destroyed.  Isocyanate resistant materials include Teflon, Viton, nitrile rubber and some PVA gloves.  Protective gloves and overalls should be worn as specified in the appropriate national standard.  Contaminated garments should be removed promptly and should not be re-used until they have been decontaminated.  When handling sealed and suitably insulated cylinders wear cloth or leather gloves.
Body protection	See Other protection below
Other protection	<ul> <li>Protective overalls, closely fitted at neck and wrist.</li> <li>Eye-wash unit.</li> <li>Ensure availability of lifeline in confined spaces.</li> </ul>

### Respiratory protection

Full face respirator with supplied air.

- ▶ Cartridge respirators should never be used for emergency ingress or in areas of unknown vapour concentrations or oxygen content.
- The wearer must be warned to leave the contaminated area immediately on detecting any odours through the respirator. The odour may indicate that the mask is not functioning properly, that the vapour concentration is too high, or that the mask is not properly fitted. Because of these limitations, only restricted use of cartridge respirators is considered appropriate.
- Cartridge performance is affected by humidity. Cartridges should be changed after 2 hr of continuous use unless it is determined that the humidity is less than 75%, in which case, cartridges can be used for 4 hr. Used cartridges should be discarded daily, regardless of the length of time used
- Positive pressure, full face, air-supplied breathing apparatus should be used for work in enclosed spaces if a leak is suspected or the primary containment is to be opened (e.g. for a cylinder change)
- Air-supplied breathing apparatus is required where release of gas from primary containment is either suspected or demonstrated.

### **SECTION 9 Physical and chemical properties**

Information on basic physical	and chemical properties		
Appearance	Moisture sensitive.		
Physical state	Compressed Gas	Relative density (Water = 1)	1.2
Odour	Not Available	Partition coefficient n-octanol / water	Not Available
Odour threshold	Not Available	Auto-ignition temperature (°C)	Not Available
pH (as supplied)	Not Available	Decomposition temperature (°C)	Not Available
Melting point / freezing point (°C)	Not Available	Viscosity (cSt)	Not Available
Initial boiling point and boiling range (°C)	208	Molecular weight (g/mol)	Not Available
Flash point (°C)	>199	Taste	Not Available
Evaporation rate	Not Available	Explosive properties	Not Available
Flammability	Not Applicable	Oxidising properties	Not Available
Upper Explosive Limit (%)	Not Available	Surface Tension (dyn/cm or mN/m)	Not Available
Lower Explosive Limit (%)	Not Available	Volatile Component (%vol)	Not Available
Vapour pressure (kPa)	Not Available	Gas group	Not Available
Solubility in water	Partly miscible	pH as a solution (1%)	Not Available
Vapour density (Air = 1)	Not Available	VOC g/L	Not Available

### **SECTION 10 Stability and reactivity**

Reactivity	See section 7
Chemical stability	<ul> <li>Unstable in the presence of incompatible materials.</li> <li>Product is considered stable.</li> <li>Hazardous polymerisation will not occur.</li> <li>Presence of elevated temperatures.</li> </ul>
Possibility of hazardous reactions	See section 7
Conditions to avoid	See section 7
Incompatible materials	See section 7

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Hazardous decomposition products

See section 5

### **SECTION 11 Toxicological information**

#### Information on toxicological effects The material can cause respiratory irritation in some persons. The body's response to such irritation can cause further lung damage. Inhalation of the vapour is hazardous and may even be fatal The use of a quantity of material in an unventilated or confined space may result in increased exposure and an irritating atmosphere developing. Before starting consider control of exposure by mechanical ventilation. Inhalation of toxic gases may cause: Central Nervous System effects including depression, headache, confusion, dizziness, stupor, coma and seizures; respiratory: acute lung swellings, shortness of breath, wheezing, rapid breathing, other symptoms and respiratory arrest; Inhaled heart: collapse, irregular heartbeats and cardiac arrest; gastrointestinal: irritation, ulcers, nausea and vomiting (may be bloody), and abdominal pain. The vapour/mist may be highly irritating to the upper respiratory tract and lungs; the response may be severe enough to produce bronchitis and pulmonary oedema. Possible neurological symptoms arising from isocyanate exposure include headache, insomnia, euphoria, ataxia, anxiety neurosis, depression and paranoia. Gastrointestinal disturbances are characterised by nausea and vomiting. Inhalation of aerosols (mists, fumes), generated by the material during the course of normal handling, may produce severely toxic effects. Relatively small amounts absorbed from the lungs may prove fatal. There is strong evidence to suggest that this material can cause, if inhaled once, very serious, irreversible damage of organs. Not normally a hazard due to physical form of product. Considered an unlikely route of entry in commercial/industrial environments Ingestion Accidental ingestion of the material may be seriously damaging to the health of the individual; animal experiments indicate that ingestion of less than 40 gram may be fatal. This material can cause inflammation of the skin on contact in some persons. The material may accentuate any pre-existing dermatitis condition Skin contact with the material may damage the health of the individual; systemic effects may result following absorption. Skin Contact

Open cuts, abraded or irritated skin should not be exposed to this material

Entry into the blood-stream, through, for example, cuts, abrasions or lesions, may produce systemic injury with harmful effects. Examine the skin prior to the use of the material and ensure that any external damage is suitably protected.

### Eye

Not considered to be a risk because of the extreme volatility of the gas.

This material may produce eye irritation in some persons and produce eye damage 24 hours or more after instillation. Moderate inflammation may be expected with redness; conjunctivitis may occur with prolonged exposure.

There has been concern that this material can cause cancer or mutations, but there is not enough data to make an assessment. Long-term exposure to respiratory irritants may result in airways disease, involving difficulty breathing and related whole-body problems. Inhaling this product is more likely to cause a sensitisation reaction in some persons compared to the general population. Skin contact with the material is more likely to cause a sensitisation reaction in some persons compared to the general population. Toxic: danger of serious damage to health by prolonged exposure through inhalation, in contact with skin and if swallowed.

This material can cause serious damage if one is exposed to it for long periods. It can be assumed that it contains a substance which can produce severe defects. This product contains a polymer with a functional group considered to be of high concern. Isothiocyanates may cause hypersensitivity of the skin

and airways.

### Chronic

The reactivity of an epoxide intermediate may be the reason for the cancer-causing properties of halogenated oxiranes. It is reported that 1.1-dichloroethyne, vinyl chloride, trichloroethylene, tetrachloroethylene and chloroprene all cause cancer.

Generally speaking, substances with one halogen substitution show higher potential to cause cancer compared to substances with two. Main route of exposure to the gas in the workplace is by inhalation.

Persons with a history of asthma or other respiratory problems or are known to be sensitised, should not be engaged in any work involving the handling of isocvanates.

The chemistry of reaction of isocyanates, as evidenced by MDI, in biological milieu is such that in the event of a true exposure of small MDI doses to the mouth, reactions will commence at once with biological macromolecules in the buccal region and will continue along the digestive tract prior to reaching the stomach. Reaction products will be a variety of polyureas and macromolecular conjugates with for example mucus, proteins and cell components.

Animal testing shows that polymeric MDI can damage the nasal cavities and lungs, causing inflammation and increased cell growth.

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TOXICITY	IRRITATION
Not Available	Not Available

### 4,4'-diphenylmethane diisocvanate (MDI)

TOXICITY	IRRITATION
Dermal (rabbit) LD50: >6200 mg/kg <sup>[2]</sup>	Dermal Sensitiser *
Inhalation(Rat) LC50: 0.368 mg/L4h <sup>[1]</sup>	Eye: no adverse effect observed (not irritating) <sup>[1]</sup>
Oral (Rat) LD50; >2000 mg/kg <sup>[1]</sup>	Skin (rabbit): 500 mg /24 hours
	Skin: adverse effect observed (irritating) <sup>[1]</sup>

### polymeric diphenylmethane diisocyanate

TOXICITY	IRRITATION	
Dermal (rabbit) LD50: >9400 mg/kg <sup>[2]</sup>	Eye (rabbit): 100 mg - mild	
Inhalation(Rat) LC50: 0.49 mg/L4h <sup>[2]</sup>		
Oral (Rat) LD50; 43000 mg/kg <sup>[2]</sup>		

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	TOXICITY		IRRITATION
1,3,3,3-tetrafluoropropene	Inhalation(Rat) LC50: >1157.752 ppm4h <sup>[2]</sup>	Not Available	
nitrogen	TOXICITY	IRRITATION	
	Not Available	Not Available	
Legend:	Value obtained from Europe ECHA Registered Sulspecified data extracted from RTECS - Register of Total	•	obtained from manufacturer's SDS. Unless otherwise
4,4'-DIPHENYLMETHANE DIISOCYANATE (MDI)	Inhalation (human) TCLo: 0.13 ppm/30 mins Eye (rab	obit): 0.10 mg moderate	
POLYMERIC DIPHENYLMETHANE DIISOCYANATE	product		
1,3,3,3- TETRAFLUOROPROPENE	practically non-toxic. Short-term exposures at levels I toxic effects. Rats and rabbits did not exhibit any seri HFO-1234ze. Based on a series of mutagenicity and observed in dogs with exposures up to 120,000 ppm; vitro genotoxicity findings include negative Ames Test in the mouse micronucleus test were negative (inhala	nigher than 10% have not induced ous toxic, developmental or reproc genomics studies, the cancer risk repeated dose toxicity in rats (13- t and negative human lymphocyte ution, mammalian bone-marrow cyl- kidney damage, brain changes and	for HFO-1234ze is low, no cardiac sensitisation was wk) found mild effects on the heart (NOEL 5,000ppm); in chromosome aberration test; in vivo genotoxicity findings togenic test with chromosomal analysis). It death. Repeated exposures may alter blood pressure
NITROGEN	No significant acute toxicological data identified in lite	erature search.	
HandiFoam FR HFO & HandiFoam FR Black HFO A-side & 4,4'-DIPHENYLMETHANE DIISOCYANATE (MDI) & POLYMERIC DIPHENYLMETHANE DIISOCYANATE	known as reactive airways dysfunction syndrome (RA criteria for diagnosing RADS include the absence of pasthma-like symptoms within minutes to hours of a dot Allergic reactions involving the respiratory tract are us potential of the allergen and period of exposure often others, and exposure to other irritants may aggravate Attention should be paid to atopic diathesis, characte Exogenous allergic alveolitis is induced essentially by lymphocytes) may be involved. Such allergy is of the The following information refers to contact allergens a	ADS) which can occur after exposu- previous airways disease in a non- procumented exposure to the irritant sually due to interactions between determine the severity of symptor a symptoms. I rised by increased susceptibility to a allergen specific immune-comple delayed type with onset up to four as a group and may not be specific tact eczema, more rarely as urtica	atopic individual, with sudden onset of persistent  IgE antibodies and allergens and occur rapidly. Allergic ns. Some people may be genetically more prone than  nasal inflammation, asthma and eczema.  xes of the IgG type; cell-mediated reactions (T hours following exposure.
HandiFoam FR HFO & HandiFoam FR Black HFO A-side & 1,3,3,3- TETRAFLUOROPROPENE	Disinfection byproducts (DBPs) are formed when disinfectants such as chlorine, chloramines and ozone react with organic and inorganic matter in water. Animal studies have shown that some DBPs cause cancer. To date, several hundred DBPs have been identified.  Numerous haloalkanes and haloalkenes have been tested for cancer-causing and mutation-causing activities.		
4,4'-DIPHENYLMETHANE DIISOCYANATE (MDI) & POLYMERIC DIPHENYLMETHANE DIISOCYANATE	The substance is classified by IARC as Group 3:  NOT classifiable as to its carcinogenicity to humans.  Evidence of carcinogenicity may be inadequate or limited in animal testing.  Isocyanate vapours are irritating to the airways and can cause their inflammation, with wheezing, gasping, severe distress, even loss of consciousness and fluid in the lungs. Nervous system symptoms that may occur include headache, sleep disturbance, euphoria, inco-ordination, anxiety, depression and paranoia.  The material may produce moderate eye irritation leading to inflammation. Repeated or prolonged exposure to irritants may produce conjunctivitis.  Aromatic and aliphatic diisocyanates may cause airway toxicity and skin sensitization. Monomers and prepolymers exhibit similar respiratory effect. Of the several members of diisocyanates tested on experimental animals by inhalation and oral exposure, some caused cancer while others produced a harmless outcome.		
Acute Toxicity	<b>✓</b>	Carcinogenici	ty 🗸
Skin Irritation/Corrosion	·	Reproductivi	•
Serious Eye Damage/Irritation	<b>→</b>	STOT - Single Exposu	
Respiratory or Skin sensitisation	<b>✓</b>	STOT - Repeated Exposu	
Mutagenicity	×	Aspiration Haza	rd X
-	1	·	er not available or does not fill the criteria for classification

Data available to make classification

### **SECTION 12 Ecological information**

## Toxicity

HandiFoam FR HFO & HandiFoam FR Black HFO	Endpoint	Test Duration (hr)	s	Species	Value	Source
A-side	Not Available	Not Available	N	Not Available	Not Available	Not Available
		<u>'</u>	<u> </u>	·		
	Endpoint	Test Duration (hr)	Species		Value	Source

4,4'-diphenylmethane		
diisocyanate (MDI)		

Endpoint	Test Duration (hr)	Species	Value	Source
EC50	72h	Algae or other aquatic plants	>1640mg/l	2
BCF	672h	Fish	61-150	7

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	NOEC(ECx) 504h		Crustacea			>=10mg/l	2	2		
	LC50	96h		Fish	Fish		95.24-134.37mg/l		Not Available	
					Species					
polymeric diphenylmethane	Endpoint		Test Duration (hr)	Test Duration (hr)		Va	Value		Source	
diisocyanate	Not Available		Not Available		Not Available	No	ot Available	Not A	vailable	
	Endpoint Test Duration (hr)		Species		Valu	9	Source			
4.0.0.0.4.4	EC50	72h		Algae or other aquatic plants		>170	mg/l	2		
1,3,3,3-tetrafluoropropene	EC50	48h		Crustacea		>160	mg/l	2		
	EC50(ECx)	EC50(ECx) 48h		Crustacea >1		>160	mg/l	2		
nitrogen	Endpoint Test Duration (hr)		Test Duration (hr)		Species V		Value		ce	
	Not Available Not Available		Not Available Not Ava		Available Not Available		voilable			

Ecotox database - Aquatic Toxicity Data 5. ECETOC Aquatic Hazard Assessment Data 6. NITE (Japan) - Bioconcentration Data 7. METI (Japan) - Bioconcentration Data 8. Vendor Data

for polyisocyanates:

Polyisocyanates are not readily biodegradable. However, due to other elimination mechanisms (hydrolysis, adsorption), long retention times in water are not to be expected. The resulting polyurea is more or less inert and, due to its molecular size, not bioavailable.

For Isocyanate Monomers:

Environmental Fate: Isocyanates, (di- and polyfunctional isocyanates), are commonly used to make various polymers, such as polyurethanes. Polyurethanes find significant application in the manufacture of rigid and flexible foams. They are also used in the production of adhesives, elastomers, and coatings. DO NOT discharge into sewer or waterways.

### Persistence and degradability

Ingredient	Persistence: Water/Soil	Persistence: Air
4,4'-diphenylmethane diisocyanate (MDI)	LOW (Half-life = 1 days)	LOW (Half-life = 0.24 days)

### **Bioaccumulative potential**

Ingredient	Bioaccumulation
4,4'-diphenylmethane diisocyanate (MDI)	LOW (BCF = 15)

### Mobility in soil

Ingredient	Mobility
4,4'-diphenylmethane diisocyanate (MDI)	LOW (KOC = 376200)

### **SECTION 13 Disposal considerations**

### Waste treatment methods

Product / Packaging disposal

- ▶ DO NOT allow wash water from cleaning or process equipment to enter drains.
- It may be necessary to collect all wash water for treatment before disposal.
- In all cases disposal to sewer may be subject to local laws and regulations and these should be considered first.
- ▶ Evaporate residue at an approved site.
- ▶ Return empty containers to supplier. If containers are marked non-returnable establish means of disposal with manufacturer prior to purchase.

### **SECTION 14 Transport information**

### **Labels Required**



NO

**Marine Pollutant** 

### Land transport (DOT)

UN number	3500		
UN proper shipping name Chemical under pressure, n.o.s. (Hydrofluoroolefin, Nitrogen)			

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Transment beyond along (as)	Class	2.2						
Transport hazard class(es)	Subrisk	Not Appl	icable					
Packing group	Not Applicable							
Environmental hazard	Not Applicable							
Special precautions for user	Hazard Lab	bel	2.2					
opecial precautions for user	Special pro	ovisions	362, T50, TP40					

### Air transport (ICAO-IATA / DGR)

All transport (ICAO-IATA / DGF	<u>'</u>					
UN number	3500	3500				
UN proper shipping name	Chemical under pressure	e, n.o.s. * (Hydrofluoroolefin, Nitrogen)				
Transport hazard class(es)	ICAO/IATA Class 2.2 ICAO / IATA Subrisk Not Applicable ERG Code 2L					
Packing group	Not Applicable	Not Applicable				
Environmental hazard	Not Applicable					
Special precautions for user		Qty / Pack Packing Instructions	A187 218 150 kg 218 75 kg Forbidden			

### Sea transport (IMDG-Code / GGVSee)

UN number	3500	3500				
UN proper shipping name	CHEMICAL UNDER PR	CHEMICAL UNDER PRESSURE, N.O.S. (Hydrofluoroolefin, Nitrogen)				
Transport hazard class(es)						
Packing group	Not Applicable					
Environmental hazard	Not Applicable					
Special precautions for user	EMS Number Special provisions Limited Quantities	F-C, S-V 274 362 0				

### Transport in bulk according to Annex II of MARPOL and the IBC code

Not Applicable

### Transport in bulk in accordance with MARPOL Annex V and the IMSBC Code

Product name	Group
4,4'-diphenylmethane diisocyanate (MDI)	Not Available
polymeric diphenylmethane diisocyanate	Not Available
1,3,3,3-tetrafluoropropene	Not Available
nitrogen	Not Available

### Transport in bulk in accordance with the ICG Code

Product name	Ship Type
4,4'-diphenylmethane diisocyanate (MDI)	Not Available
polymeric diphenylmethane diisocyanate	Not Available
1,3,3,3-tetrafluoropropene	Not Available
nitrogen	Not Available

### **SECTION 15 Regulatory information**

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#### 4,4'-diphenylmethane diisocyanate (MDI) is found on the following regulatory lists

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US - California Hazardous Air Pollutants Identified as Toxic Air Contaminants

US - Massachusetts - Right To Know Listed Chemicals

US Clean Air Act - Hazardous Air Pollutants

US DOE Temporary Emergency Exposure Limits (TEELs)

US EPA Integrated Risk Information System (IRIS)

### polymeric diphenylmethane diisocyanate is found on the following regulatory lists

International Agency for Research on Cancer (IARC) - Agents Classified by the IARC Monographs

US ATSDR Minimal Risk Levels for Hazardous Substances (MRLs)

US DOE Temporary Emergency Exposure Limits (TEELs)

### 1,3,3,3-tetrafluoropropene is found on the following regulatory lists

US AIHA Workplace Environmental Exposure Levels (WEELs)

US DOE Temporary Emergency Exposure Limits (TEELs)

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

### nitrogen is found on the following regulatory lists

US - Massachusetts - Right To Know Listed Chemicals

US DOE Temporary Emergency Exposure Limits (TEELs)

US EPCRA Section 313 Chemical List

US NIOSH Recommended Exposure Limits (RELs)

US OSHA Permissible Exposure Limits (PELs) Table Z-1

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US TSCA Chemical Substance Inventory - Interim List of Active Substances US TSCA New Chemical Exposure Limits (NCEL)

US EPCRA Section 313 Chemical List
US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory

US TSCA Chemical Substance Inventory - Interim List of Active Substances

US Toxicology Excellence for Risk Assessment (TERA) Workplace Environmental Exposure Levels (WEEL)

US TSCA Chemical Substance Inventory - Interim List of Active Substances

US Toxic Substances Control Act (TSCA) - Chemical Substance Inventory US TSCA Chemical Substance Inventory - Interim List of Active Substances

### **Federal Regulations**

### Superfund Amendments and Reauthorization Act of 1986 (SARA)

# Section 311/312 hazard categories

Flammable (Gases, Aerosols, Liquids, or Solids)	No
Gas under pressure	Yes
Explosive	No
Self-heating Self-heating	No
Pyrophoric (Liquid or Solid)	No
Pyrophoric Gas	No
Corrosive to metal	No
Oxidizer (Liquid, Solid or Gas)	No
Organic Peroxide	No
Self-reactive	No
In contact with water emits flammable gas	No
Combustible Dust	No
Carcinogenicity	Yes
Acute toxicity (any route of exposure)	Yes
Reproductive toxicity	No
Skin Corrosion or Irritation	Yes
Respiratory or Skin Sensitization	Yes
Serious eye damage or eye irritation	Yes
Specific target organ toxicity (single or repeated exposure)	Yes
Aspiration Hazard	No
Germ cell mutagenicity	No
Simple Asphyxiant	No
Hazards Not Otherwise Classified	No

### US. EPA CERCLA Hazardous Substances and Reportable Quantities (40 CFR 302.4)

Name	Reportable Quantity in Pounds (lb)	Reportable Quantity in kg
4,4'-diphenylmethane diisocyanate (MDI)	5000	2270

### State Regulations

### US. California Proposition 65

None Reported

### **National Inventory Status**

National Inventory	Status
Australia - AIIC / Australia Non-Industrial Use	Yes
Canada - DSL	Yes
Canada - NDSL	No
China - IECSC	Yes

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National Inventory	Status	
Europe - EINEC / ELINCS / NLP	Yes	
Japan - ENCS	Yes	
Korea - KECI	Yes	
New Zealand - NZIoC	No (1,3,3,3-tetrafluoropropene)	
Philippines - PICCS	No (1,3,3,3-tetrafluoropropene)	
USA - TSCA	Yes	
Taiwan - TCSI	Yes	
Mexico - INSQ	No (1,3,3,3-tetrafluoropropene)	
Vietnam - NCI	Yes	
Russia - FBEPH	Yes	
Legend:	Yes = All CAS declared ingredients are on the inventory No = One or more of the CAS listed ingredients are not on the inventory. These ingredients may be exempt or will require registration.	

### **SECTION 16 Other information**

Revision Date	11/09/2022
Initial Date	02/28/2021

### CONTACT POINT

### **SDS Version Summary**

Version	Date of Update	Sections Updated
1.4	11/09/2022	Physical Properties, Name

### Other information

Classification of the preparation and its individual components has drawn on official and authoritative sources as well as independent review by the Chemwatch Classification committee using available literature references

committee using available literature references.

The SDS is a Hazard Communication tool and should be used to assist in the Risk Assessment. Many factors determine whether the reported Hazards are Risks in the workplace or other settings. Risks may be determined by reference to Exposures Scenarios.

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